



A few silicate minerals



Nesosilicates / Garnets $(X_3^{2+}Y_2^{3+}[SiO_4]_3)$

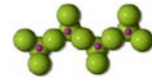


Use: gemstone, abrasive



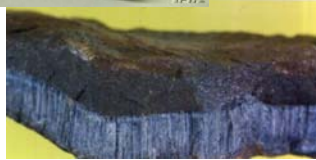


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Inosilicates / Amphiboles

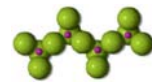
Bleu asbestos: Crocidolite ($\text{Na}_2(\text{Fe,Mg})_5\text{Si}_8\text{O}_{22}(\text{OH})_2$)



Sony materials and conservation of the built heritage – Natural Stone – Mineralogy – BR/20

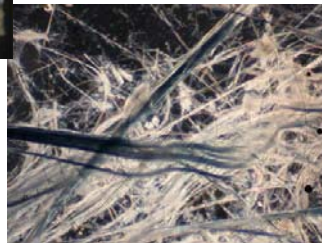
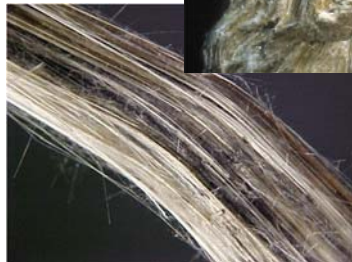


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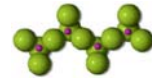


Inosilicates / Amphiboles

Brown asbestos: Amosite ($\text{Fe}_7\text{Si}_8\text{O}_{22}(\text{OH})_2$)

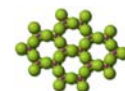
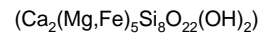


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Inosilicates / Amphiboles

Green asbestos. ex. : Actinolite



Phyllosilicates / Micas

Muscovite $(KAl_2[(OH,F)_2AlSi_3O_{10}])$ white mica,

Biotite $(K(Mg,Fe^{2+},Mn^{2+})_3[(OH,F)_2(Al,Fe^{3+},Ti^{3+})Si_3O_{10}])$ black/brown mica

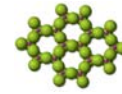
Use: Heat, acoustic and electric insulator, paints



muscovite



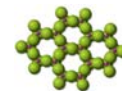
biotite



Phyllosilicates / Clay minerals

Kaolinite $(Al_2Si_2O_5(OH)_4)$

Use:
porcelain
manufacture,
filler in papers



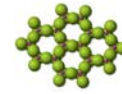
Phyllosilicates / Clay minerals

Kaolinite $(Al_2Si_2O_5(OH)_4)$

TABLE 2. Industrial uses of kaolin.

Paper coating	Cement	Food additives
Paper filling	Pencil leads	Bleaching
Extender in paint	Adhesives	Fertilizers
Ceramic raw material	Tanning leather	Plaster
Filler in rubber	Pharmaceuticals	Filter aids
Filler in plastics	Enamels	Cosmetics
Extender in ink	Pastes and glues	Crayons
Cracking catalysts	Insecticide carriers	Detergents
Fibreglass	Medicines	Roofing granules
Foundries	Sizing	Linoleum
Desiccants	Textiles	Polishing compounds

H. H. Murray, 1999. Applied clay mineralogy today and tomorrow, Clay Minerals, V.34, p. 39-49

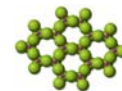


Phyllosilicates / Clay minerals

Montmorillonite $((\text{Na,Ca})_{0.3}(\text{Al,Mg})_2\text{Si}_4\text{O}_{10}(\text{OH})_2 \cdot n\text{H}_2\text{O})$

Swelling clay mineral exchanger of ions

Use: gastric plaster, cleaner of greases (Terre de Sommières), bentonite, container for the nuclear waste



Phyllosilicates / Clay minerals

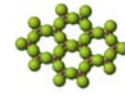
Montmorillonite $((\text{Na,Ca})_{0.3}(\text{Al,Mg})_2\text{Si}_4\text{O}_{10}(\text{OH})_2 \cdot n\text{H}_2\text{O})$

= smectite = bentonite

TABLE 3. Industrial uses of smectites.

Drilling mud	Medical formulations	Crayons
Foundry bond clay	Polishing & cleaning agents	Cement
Pelletizing iron ores	Detergents	Desiccants
Sealants	Aerosols	Cosmetics
Animal feed bonds	Adhesives	Paint
Bleaching clay	Pharmaceuticals	Paper
Industrial oil absorbents	Food additives	Fillers
Agricultural carriers	De-inking of paper	Ceramics
Cat box absorbents	Tape-joining compounds	Catalysts
Beer and wine clarification	Emulsion stabilizer	Pencil leads

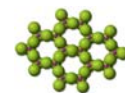
H. H. Murray, 1999. *Applied clay mineralogy today and tomorrow*, Clay Minerals, V.34, p. 39-49



Phyllosilicates / Clay minerals

Talc ($Mg_3Si_4O_{10}(OH)_2$)

Use of talc:
cosmetic,
lubricant,
manufacture of
paper, excipient
and lubricant in
the
pharmaceutical
industry, tailor's
chalk

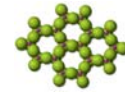


Phyllosilicates / Chlorites

$((Mg,Fe,Mn,Al)_6((Si,Al)_4O_{10})(OH)_8)$

Use:
decorative stone





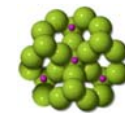
Phyllosilicates / Serpentinites

White asbestos: Chrysotile ($Mg_3Si_2O_5(OH)_4$)

Use: reinforced cement, machine parts under friction, joints for high temperature machines... because non flammable, imputrescible, flexible, resistant to the majority of chemicals and with a high breaking stress => majority of the world market of asbestos



<http://www.ec.gc.ca/nopp/docs/consult/Rotterdam/ca/fr/chrysotileBG.cfm>



Tectosilicates / Quartz

(SiO_2)

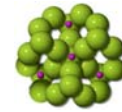
Use:
Piezoelectric
(clock industry, ...)
and...



citrine



gemstones (amethyst, citrine)...
cryptocrystalline varieties: flint,
agate, onyx, carnelian,
jasper, opal



Tectosilicates / Feldspars

K-feldspars (KAISi_3O_8)



microcline



amazonite



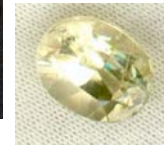
orthose



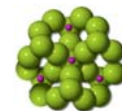
microcline



sanidine



Use:
ceramics,
porcelain, glass,
bricks, soaps,
scouring
powders,
gemstones



Tectosilicates / Feldspars

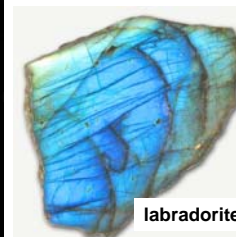
Plagioclases (AlSi_3O_8)(Ca,Na)



anorthite



albite



labradorite



albite

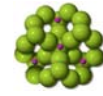


oligoclase

Use:
ceramics,
porcelain, glass,
bricks, soaps,
scouring
powders,
gemstones



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Tectosilicates / Feldspathoids

Lazurite $((\text{Na}, \text{Ca})_6(\text{Al}, \text{Si})_{12}\text{O}_{24}\text{S}_2 \text{FeS} \cdot \text{CaCO}_3)$



Use: gemstone, blue pigment

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A few non silicate minerals

Sony materials and conservation of the built heritage – Natural Stone – Mineralogy – BR /35



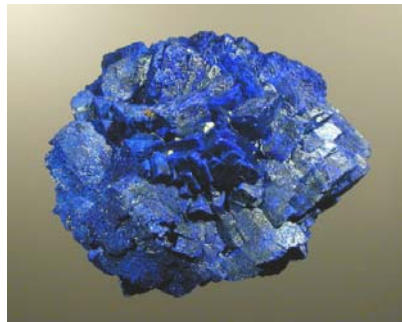
Carbonates / Calcite (CaCO_3)

Use: white pigment (calcite as chalk used since prehistory), raw material of lime



Carbonates / Azurite ($2\text{CuCO}_3 \cdot \text{Cu}(\text{OH})_2$)

Use: blue pigment, gemstone





Carbonates / Malachite ($\text{CuCO}_3, \text{Cu(OH)}_2$)

Use: green pigment,
gemstone



polished roller



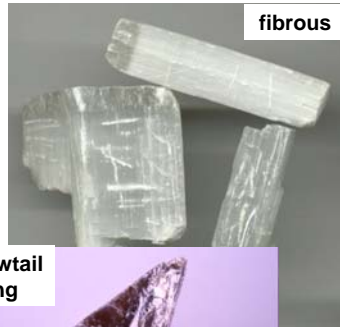
Carbonates / Cerussite (PbCO_3)

Use: cosmetic
(in the past
since antiquity);
white pigment
(= white lead)





Sulfates / Gypsum ($\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$)



Selenite (= pierre de lune)



Sulfates / Gypsum ($\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$)

Use: raw material of plaster; fertilizer and soil conditioner, Tofu coagulant, blackboard chalk





Sulfates / Gypsum alabaster ($\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$)

Use: decorative stone



Sulfates / Gypsum ($\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$)

Weathering mineral: coming from the air pollution, the stone itself or from cements => degrading stones





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Sulfates / Barite (BaSO_4)

Use: major source of barium, white pigment (blanc fixe), used in paper or paint manufacturing, radiography, heavy filler



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Halides / Halite (NaCl)

Use: table salt, road salt

Danger for building stones:
crystallisation
damp patches



Sony materials and conservation of the built heritage – Natural Stone – Mineralogy – BR /45



Halides / Sylvite (KCl)

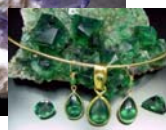
Use: fertilizer, substitute for table salt, lethal injection

Danger for building stones: crystallisation, damp patches



Halides / Fluorite (CaF₂)

Use: manufacture of hydrofluoric acid, enamels, glass fibre; used as camera lens; purple pigment; gemstone

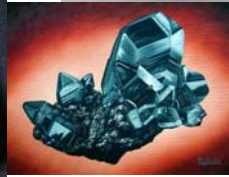




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Oxides / Hematite (Fe_2O_3)

Use: red pigment;
gemstone



Sony materials and conservation of the built heritage – Natural Stone – Mineralogy – BR 4/8



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Oxides / Goethite ($\text{FeO}(\text{OH})$)

Use: yellow pigment



Sony materials and conservation of the built heritage – Natural Stone – Mineralogy – BR 4/9



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Oxides / Rutile (TiO_2)

Use: white pigment (artificial); manufacture of paints; +/- in gemstones



Sony materials and conservation of the built heritage – Natural Stone – Mineralogy – BR/50



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Oxides / Corundum (Al_2O_3)

Use: abrasive; gemstones

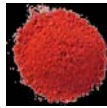


Sony materials and conservation of the built heritage – Natural Stone – Mineralogy – BR/51



Oxides / Minium (Pb_3O_4)

Use: red pigment, manufacture of glass, protecting paint against the corrosion of metals



Oxides / Massicot (or litharge) (PbO)

Use: yellow pigment, manufacture of glass, of oils and varnishes (desiccant), production of insecticides





Sulfides / Galena (PbS)

Use: black pigment, cosmetic (khol), semiconductor in old wireless systems



Sulfides / Pyrite (FeS₂) (= fool's gold)

Use: production of sulfur dioxide for paper industry or manufacturing of sulfuric acid

"Dangers" in building oxidation makes it dangerous in aggregates of concrete; rost patches on stones (marble, sandstones,...)





Sulfides / Realgar (AsS)

Use: red pigment;
fireworks

Problems: unstable
with light (\Rightarrow yellow
pararealgar)



Sulfides / Orpiment (As₂S₃)

Use: yellow pigment;
production of
semiconductors and
photoconductors,
fireworks

Problems:
incompatible with
pigments like lead
and copper-based; it
blackens in contact
with the air





Sulfides / Cinnabar (HgS)

Use: red pigment; medicine, drug, food dye

Problems: it blackens in contact with the air



Phosphates / Apatite ($\text{Ca}_5(\text{PO}_4)_3(\text{OH}, \text{F}, \text{Cl})$)

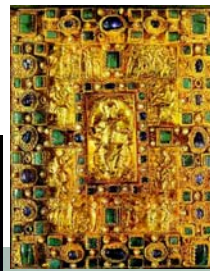
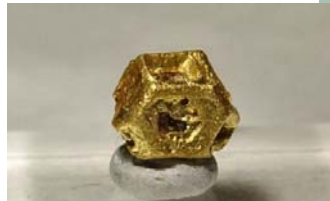
Use: fertilizer; gemstone; new stone consolidant





Elements / Gold (Au)

Use: noble metal, decorative metal, gilding; conductive coating, money



Elements / Silver (Ag)

Use: noble metal; decorative metal; printed circuits; electrical contacts; dental alloys; antibacterial; money

